
Still looking for good resources to review. Thanks. If you are still looking for resources, you can try the following books: Fundamentals of Soil Physics by Lindsay Upson. Fundamentals of Geophysics by Fanning Gannett, or Electricity and Magnetism by Tipler Ralph. Wanted to ask that you email me the answer sheet of the problem on the "power plant" that is on your website. I am trying to solve it and I do not know what to do. The following is a solution for Lab #7, Part B of Redox Lab #2. If you need help with any other lab, you can email me at patricio.cifuentes@gmail.com. Enjoy. This is a solution for Lab #7, Part A of Redox Lab #2. If you need help with any other lab, you can email me at patricio.cifuentes@gmail.com. Enjoy. If anyone needs help with this problem, you can email me at patricio.cifuentes@gmail.com and I'll be happy to help you out! Good luck!!

Solution: (a) Consider the following reaction: $[Oxygen + 2 HCl] \rightarrow [HHO + CO]$. First, we can write the oxidation numbers for this reaction: $[Oxygen] + 2 HCl \rightarrow [HHO + CO]$. Since O is an element that does not have any oxidation number, there are 6 entries in the 6x6 table. The first entry is the element's oxidation number, then three hyphens are placed to denote that there are six entries in the 6x6 table. Therefore, the first entry is $6/3 = 2/3$, so there are two sets of three entries $((2/3), (2/3), (2/3))$. The O entry in these sets have three hyphens placed after them to indicate that there are six entries in the 6x6 table. The 3-hyphen entry means that there is a two times a two times a two, or twelve entries $((2/3), (2/3), (2/3), (2/3), (2/3), (2/3), (2/3), (2/3), (2/3), (2/3))$. The HHOs in the above reaction have an oxidation number of +4. Therefore, they are in the fourth row in each set. The CO in HHO has an oxidation number of -1, thus, it is the eleventh entry from the bottom in each set.

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